



Toxic metals adsorption from water by recycling waste product

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Outline



- 2. Scope of the study
- 3. Methodology
- 4. Results and Discussion
- 5. Conclusions & Recommendation









INTRODUCTION

- Mining, ore smelting, leather tanning, pulp industry, painting, pesticide, and pharmaceutical manufacturing industries release metal contaminants
- Toxic metals such as lead (Pb), Copper (Cu), Cadmium (Cd), Zinc (Zn), Arsenic (As), Chromium (Cr), Cobalt (Co), and Nickle (Ni) etc.
- Many of the heavy metals are well known to be a mutagen causing cancer









INTRODUCTION











Fig: Industrial discharges



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INTRODUCTION

<u>Table 1</u>: Heavy metal concentration (ppm) of the studied wastewater samples

	[I]							
	Samples	Ni	Mn	Cr	Zn	Cd	Pb	Cu
	1	5.36	1	0.52	1.44	0.84	3.2	0.52
	2	4.36	1.88	0.72	1.4	1.4	10	2.24
	3	1.6	1.84	1	1.44	1.92	5.2	2.4
	4	3.28	0.88	0.92	2	0.8	12.8	1.32
	5	9	0.52	1.24	1.68	1.04	4.8	1.4
EA	Saudi Standard (MWE, 2006)	5	0.2	0.1	2	0.01	0.1	0.4
	WHO Standard (2006)	10	1	1	5	0.1	0.1	1



Heavy metal contamination in the coastal region of Jeddah City





INTRODUCTION







KSA Mining:

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- Gold and Base Metals,
 - Phosphate,
 - Aluminium,
- Industrial Minerals including magnesite, kaolin and low-grade bauxite

Figures : Slag production from metal mining site





SCOPE OF THE STUDY

- The use of nickel smelter slags (e.g. an iron oxide and iron silicate bearing waste from Ni-ore smelting) in Cu, Pb, Cd and Zn removal by adsorption
- To experimentally investigate the adsorption capacity and rate of adsorption of metal species by smelter waste product in the laboratory batch test, and assess the potential utilization of the slag as a reactive medium in the treatment of metal contaminated water.
- Determined the kinetics of Copper (Cu), lead (Pb), Zinc (Zn) and Cadmium (Cd) uptake by recycling slag
- Explored the benefits of using renewable waste products for water and wastewater treatment industry







METHODOLOGY

Reactive slag waste product

• Nickel smelter slag, an industrial waste, was obtained from Vale Canada Copper Cliff Smelter, Sudbury, Ontario





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Methodology

Reactive slag waste product



- The components of the bulk slag were also analyzed using PANanalytical PW-2400 Wavelength Dispersive XRF (X-Ray Fluorescence) after roasting 1 g of slag to loss of ignition
- Hitachi SU6600 Field Emission Scanning Electron Microscope (FESEM) was also used to scan the surface of the samples.
- The changes of surface charge on reactive media with different pHs were determined by Surpass Anton Parr Analyzer







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Methodology

Table 1: Composition of Ni smelter slag and other slags

Major Constituent	Iron/Blast furnace slag [12]	Steel Slag [12]	Cu smelter Slag [11]	Ni smelter Slag (the present study)
FeO	0.45%	42-84%	39%	57.3%
SiO ₂	40%	18%	31%	39.7%
CaÓ	30-40%	25-55%	3.95%	1.27%
Al ₂ O ₃	15%	3%	2.4%	2.45%
MgO	0.1%	0.57%	2.82%	1.23%
TiO ₂			0.19%	0.23%







Methodology

The changes of surface charge [mV] with pH







Figure 1: Electro kinetic Analysis of Nickel smelter slag showing pzc using the Surpass Anton Parr Analyser





Methodology





Figure 3: XRD patterns showing fresh Ni smelter slag









Methodology

Batch Test

Batch experiment was conducted by shaking known amounts of nickel slag with certain mL metal solutions in 125-mL Erlenmeyer flask at room temperature (23°C \pm 2°C). This test measured the adsorption efficiency of smelter slag.

All experiments were carried out in triplicate on the rotary shaker (New Jersey, USA) at a shaking speed of 170 rpm

- Effect of particle size
- Effect of other ions
- Equilibrium adsorption tests
- Desorption tests





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RESULTS



Figure 5: The effect of pH on the removal of Cu and Pb by nickel smelter slag (initial concentration: 10 mg/L, slag dose: 10 g/L, contact time: 10 hours, shaking speed: 170 rpm, temperature: 25^oC). standard deviation: ±0.601; standard error: ±0.36





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Results



Figure 6: The effect of pH on the removal of Cu, Cd, Ni, Pb and Zn from mixed solutions. (Initial concentration for all metals: 10 mg/L; except for Cd concentration: 7 mg/L; slag dose: 10 g/L, contact time: 10 hours, shaking speed: 170 rpm, temperature: 25° C, pH= 5; standard deviation: ± 0.432 ; standard error: ± 0.436)





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Results



Figure 7: Effect of slag dose on removal of Pb or Cu by nickel smelter slag (Initial concentration: 10 mg/L, contact time: 10 hours, shaking speed: 170 rpm, temperature: 25^oC)





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Results



Figure 8: Effect of slag sizes on removal of Pb or Cu by nickel smelter slag (Initial concentration: 10 mg/L, contact time: 10 hours, shaking speed: 170 rpm, temperature: 25^oC)





Results







Figure 9: Effect of multi cations on removal of Pb or Cu by nickel smelter slag (Initial concentrations, (Cu,Ni, Pb, and Zn): 10 mg/L and Cd = 7 mg/L, contact time: 10 hours, shaking speed: 170 rpm, temperature: 25^oC)





(a) Pb

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b) Cu

Figure 10: Langmuir isotherm plot for adsorption of (a) Pb and (b) Cu by nickel smelter slag (initial concentration: 10-100 mg/L, slag dose: 10 g/L, contact time: 24 hours, shaking speed: 170 rpm, temperature: 25^o C)







Figure 11: Removal of (a) Pb, (b) Cu and (C) mixed metals by nickel smelter slag in 5 cycles of adsorption (initial concentration: 7-10 mg/L, slag dose: 80 g/L, contact time: 10 hours, shaking speed: 170 rpm, temperature: 25° C)





CONCLUSIONS & RECOMMENDATION

- The only major cost related to the use of recycled Ni smelter slag in a reactive permeable barrier would be that of crushing.
- Low-cost, effective technologies and inexpensive treatment materials that are readily available and easily reusable offer attractive options in engineering applications..
- However, in comparison to the large footprint and the cost involved in slag disposal, it would seem that using crushed slag for environmental remediation is advantageous, both from the economic and sustainability points of view.





Conclusions & Recommendation

- The development of green technology using recycled or reused waste products as well as proposing the methods how to reduce the amount of industrial waste for disposal.
- Slag recycling can reduce or eliminate the cost of final disposal as well as environmental contamination.
- Application of reactive slag for In Situ site remediation technology





















THANK YOU





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